First-Principles Electronic Structure of Solid Picene

Taichi Kosugi^{1,2}, Takashi Miyake^{2,4}, Shoji Ishibashi², Ryotaro Arita^{3,4} and Hideo Aoki¹

Department of Physics, University of Tokyo, Hongo, Tokyo 113-0033, Japan
 Research Institute for Computational Sciences (RICS), AIST, Umezono, Tsukuba 305-8568, Japan
 Department of Applied Physics, University of Tokyo, Hongo, Tokyo 113-8656, Japan
 Japan Science and Technology Agency, CREST, Honcho, Kawaguchi, Saitama 332-0012, Japan

To explore the electronic structure of the first aromatic superconductor, potassium-doped solid picene which has been recently discovered by Mitsuhashi $et\ al.$ with the transition temperatures $T_c=7-20\ \rm K$, we have obtained a first-principles electronic structure of solid picene as a first step toward the elucidation of the mechanism of the superconductivity. The undoped crystal is found to have four conduction bands, which are characterized in terms of the maximally localized Wannier orbitals. We have revealed how the band structure reflects the stacked arrangement of molecular orbitals for both undoped and doped (K₃picene) cases, where the bands are not rigid. The Fermi surface for K₃picene is a curious composite of a warped two-dimensional surface and a three-dimensional one.

KEYWORDS: picene, superconductivity, aromatic compound, first-principles calculation, electronic structure

Introduction — While there are mounting interests toward superconductivity in more versatile classes in recent years, fascination continues with superconductivity in π electron (organic) systems. This has a long history, dating back to superconductivity in graphite intercalation compounds (GIC's), first discovered in 1965 in alkalmetal doped AC_8 (A = K, Rb, Cs) with transition temperatures $T_c < 1$ K, which was recently raised to 11.5 K in CaC₆.² The superconductivity in organic metals was first found in TMTSF compounds³ in 1980 with $T_c = 0.9$ K under 12 kbar. Then comes the BEDT-TTF family,⁴ for which $T_c = 14.2 \text{ K}$ under 82 kbar was found in 2003, the highest T_c to date among the organic superconductors. Another family is the doped fullerides, discovered⁵ in 1991 for K_3C_{60} with $T_c = 18$ K at ambient pressure, followed by Cs_2RbC_{60} with $T_c=33$ K at ambient pressure⁶ and Cs₃C₆₀ with $T_c = 40$ K under 15 kbar,⁷ which is the highest T_c among carbon-based materials. However, a new class of superconducting π -electron materials has not been found over a decade.

So a recent discovery of superconductivity by Mitsuhashi et al. 8 in a solid of aromatic molecule picene, with $T_c = 7$ and 20 K when doped with potassium, is a long-awaited breakthrough. The discovery is seminal in that this is the first aromatic superconductor, so that a comparison with other classes of π -electron superconductors should be interesting. A picene molecule consists of five benzene rings joined in an armchair manner with a higher chemical stability than pentacene with a straight arrangement. The solid picene has in fact a much larger band gap ($\simeq 3.3 \text{ eV}$)⁹ than that of solid pentacene ($\simeq 1.8 \text{ eV}$), which is why the material originally attracted attention as an organic semiconductor in e.g. organic FET's. 9, 10 Picene has been known to crystallize in orthorhombic and monoclinic forms, 11 and Kubozono's group discovered superconductivity in potassiumdoped monoclinic crystalline picene.⁸ The crystal comprises layers stacked in c direction, where each layer (in ab plane) has picene molecules arranged in a herringbone structure. The measured lattice parameters of the crystalline picene are a=8.480, b=6.154, c=13.515 in Å with $\beta=90.46^{\circ},^{11}$ where a primitive cell contains two inequivalent molecules A and B whose positions are related with a space group $P2_1$. In 8 K_xpicene with the concentration x of K atoms per picene is found to become a superconductor for $x\simeq 3$, and the Meissner effect has been detected with $H_{c2}>10^4$ Oe suggested.

Theoretically, the first thing we have to do is firstprinciples electronic structure calculations, which we have done in this Letter, with the density functional theory (DFT) for both pristine and doped solids of picene as a first step to elucidate the mechanism of superconductivity. We have identified the main characters of the conduction bands in terms of the relevant molecular orbitals, and we have also constructed a downfolded four-band tight-binding model with maximally localized Wannier orbtials. From these we shall show how the band structure reflects the stacked arrangement of molecular orbitals for both undoped picene and K₃picene. The Fermi surface for K₃picene has turned out to be a curious composite of a warped two-dimensional surface and a threedimensional one, where both the molecular arrangements modified by doping and hybridization with alkali-metal atoms are relevant.

First-principles bands — We have first optimized the atomic positions including hydrogen atoms with the lattice parameters fixed at the measured values. ¹¹ For the electronic structure calculation for the optimized structure we adopt DFT with the projector augmented-wave (PAW) method ¹² with the Quantum MAterials Simulator (QMAS) package ¹³ in the local-density approximation (LDA). The pseudo Bloch wave functions are expanded by plane waves with an energy cutoff of 40 Ry and 4^3 k-points. The angle between the planes of

the inequivalent molecules in the optimized geometry is 61°, close to the measured 58°. Each molecule is not exactly planar due to the intramolecular overcrowding, as seen in the experiment, 11 with the calculated center-ofmass vectors between two inequivalent, nearest-neighbor molecules $r_B - r_A = 0.485 a \pm 0.5 b$. In the calculated electronic band structure in Fig.1 (a), we first notice that the dispersion along the c^* axis is small compared to those along the a^*b^* plane, reflecting the layered structure. The lowest 102 bands are fully occupied, with the valence band top occurring at $c^*/2$. The conduction band bottom is also located there, giving a band gap of 2.36 eV, while the experimental value is 3.3 eV. Such an underestimation is generally seen in LDA calculations. As seen in Fig.1 (a), the band dispersions along $k_{b^*} = \pm 1/2$ are doubly degenerate, unlike the triclinic crystalline tetracene and pentacene but like the monoclinic crystalline naphthalene and anthracene. 14 The electron density, displayed in Fig.1 (c), spreads over each molecules basically uniformly, with no significant dimerization seen.

For comparison we have also calculated the electronic structures of an isolated picene molecule with the molecular geometries fixed at those optimized in the crystal to obtain the HOMO and LUMO wave functions, where the LUMO-HOMO energy gap is 2.96 eV (against 1.12 eV calculated for pentacene molecule). If we regard picene and pentacene as fragments of graphene with armchair and zigzag edges, respectively, the higher stability (i.e., larger gap) in picene is consistent with the wisdom obtained in graphene, where armchairs are much stabler.

If we go back to the crystal, valence bands are the 99th (VB4) - 102nd (VB1) bands in an energy region of width 0.77 eV separated from the lower ones by 0.26 eV. Looking at the Bloch wave functions for VB1 and VB2 at Γ , we can identify that they have mainly the character of the HOMO's of picene molecule. In each layer in the crystal, the HOMO (ϕ) of each molecule splits into bonding and antibonding ones due to the overlap with its four adjacent inequivalent molecules to form VB1 and VB2 Bloch states, respectively. We can then symbolically express them, at Γ , as $\psi_{\rm VB1} = \phi_{\rm HOMO}^{\rm A} - \phi_{\rm HOMO}^{\rm B}$ and $\psi_{\rm VB2} = \phi_{\rm HOMO}^{\rm A} + \phi_{\rm HOMO}^{\rm B}$, as confirmed in Fig. 1 (a). We also note that $\psi_{\rm VB3}$ and $\psi_{\rm VB4}$ are mainly $\phi_{\rm HOMO-1}^{\rm A,B}$ in character, so that all the four valence bands mainly derive from HOMO and HOMO-1. If we now turn to the conduction bands, they are the 103rd (CB1) - 106th (CB4) bands with a width of 0.39 eV, separated from the upper ones by 0.15 eV (Fig.1 (a)). A similar analysis shows that the Bloch states ψ_{CB1} - ψ_{CB4} mainly derive

from $\phi_{\text{LUMO(+1)}}^{\text{A,B}}$ as summarized in Fig.2 (a). Downfolding — To build a tight-binding model for the electronic structure of the pristine picene, we have constructed maximally localized Wannier orbitals (WO's)¹⁵ for the four conduction bands. The DFT calculations were performed with the Tokyo Ab initio Program Package (TAPP)¹⁶ and the Troullier-Martins norm-conserving pseudopotentials¹⁷ within LDA. The Bloch wave functions were expanded in terms of plane waves with an energy cutoff of 49 Ry. The features of the calculated band structure of the conduction bands (Fig.3)

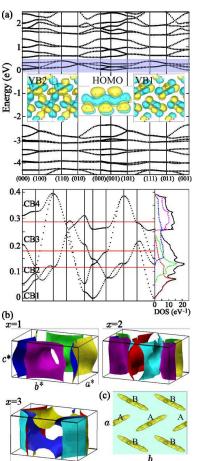


Fig. 1. (a) Calculated band structure of undoped crystalline picene. The horizontal axis is labeled by $(a^*/2, b^*/2, c^*/2)$, while the origin of the energy is set to the conduction band bottom. Insets show the highest occupied molecular orbital and the VB1 and VB2 Bloch wave functions at Γ . Bottom panel is a blowup of the conduction bands (shaded region in the top panel), with the density of states per molecule shown in the solid curve while the dashed curves the contributions from the four bands. Horizontal lines indicate E_F for x = 1, 2 and 3. (b) The Fermi surfaces when E_F is shifted to the situation in K_x picene with x = 1, 2 and 3 with rigid bands. (c) The electron density viewed along the long molecular axis.

(b)), with the atomic positions optimized above, are identical to those obtained by QMAS within the numerical accuracy. Setting the energy window to cover the four conduction bands, we have obtained four maximally localized WO's, two of which (w_{Al}, w_{Ah}) are centered at molecule A while the other two (w_{Bl}, w_{Bh}) at molecule B. The wave functions w_{Ai}, w_{Bi} (Fig.3 (a)) have the same shape around each molecule for each of i = l, h with l(h) denoting the lower (higher) energy states. The band structure in Fig.3(b) in the tight-binding model constructred from the WO's reproduces accurately the DFT-LDA band dispersion. The transfer integrals, $t_{ij}(\mathbf{d}) \equiv$ $\langle w_i(\mathbf{r})|H_{\rm KS}|w_j(\mathbf{r}-\mathbf{d})\rangle$, are defined as the matrix elements of the Kohn-Sham Hamiltonian H_{KS} between the WO's separated by $\mathbf{d} = d_1 \mathbf{a} + d_2 \mathbf{b} + d_3 \mathbf{c}$. The transfer integrals with significant magnitudes are depicted in Fig.3 (c). If we examine the shapes of w_l and w_h , they can be represented approximately as $\phi_{\text{LUMO}} \pm \phi_{\text{LUMO}+1}$, with w_h having a slightly larger Wannier spread than that of

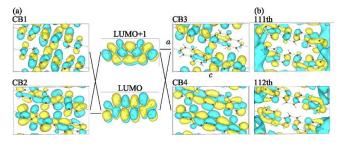


Fig. 2. (a) The lowest unoccupied molecular orbital (LUMO) and LUMO+1 for an isolated picene molecule as compared with CB1-CB4 Bloch wave functions at Γ in the crystalline picene. Solid lines indicate the main orbital components. (b) 111th and 112th Bloch wave functions at Γ that have amplitudes in the interlayer regions.

 w_l . As seen in Fig.2 (a), the LUMO (LUMO+1) are antisymmetric (symmetric) with respect to the plane that bisects the long molecular axis, and w_l and w_h are nearly mirror images of each other, with a slight polarization from the center of the molecule. The matrix element of $H_{\rm KS}$ between the LUMO and LUMO+1 in the crystal is approximately given by $|\langle \phi_{\rm LUMO}|H_{\rm KS}|\phi_{\rm LUMO+1}\rangle| \simeq (\varepsilon_h - \varepsilon_l)/2 = 11.3$ meV, where $\varepsilon_i \equiv t_{ii}(\mathbf{0})$. This is not negligible against the gap of 67.8 meV between the LUMO and LUMO+1 calculated for the molecule, which leads to the hybridization of the LUMO- and (LUMO+1)-derived bands. This contrasts with pentacene, for which the calculated gap in the molecule between the LUMO and LUMO+1 is as high as 1.28 eV, and the bands derived from them do not hybridize.

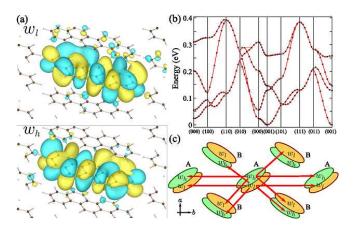


Fig. 3. (a) Maximally localized Wannier orbitals with the lower-energy w_l and the higher-energy w_h, constructed for the conduction bands. (b) Tight-binding fit (curves) as compared with the DFT-LDA band structure (dots) for the four conduction bands. (c) Arrows indicate the transfer integrals between the in-plane molecules whose magnitudes are larger than 25 meV, with ellipses representing the associated WO's, and overlaps representing the difference in the c-axis height.

Doped solid picene — Let us move on to the doped solid picene. In contrast to the case of BEDT-TTF or TMTSF, carriers are electrons in K_x picene. If we assume rigid bands and shift the Fermi energy to a position corresponding to x=3, CB3 and CB4 intersect the Fermi

level, and the Fermi surface, displayed in Fig.1 (b), shows a double surface structure. ¹⁸ However, the rigid band assumption has to be tested. The measured lattice parameters of the K_{2.9}picene are a=8.707, b=5.912, c=12.97 in Å with $\beta=92.77^\circ$. One conspicuous feature is that the c axis shrinks while the a axis expands upon doping. This contrasts with the crystalline pentacene, where significant expansions of the c axis are known to occur when doped with alkali elements a a result of the intercalants accommodated in the interlayer region. So this suggests that, in the case of picene, the dopants are not inserted into the interlayer region.

Another issure of interest is "interlayer states" in GIC. There is a body of studies, both experimental²⁰ and theoretical,²¹ that indicate the interlayer states of graphite that reside in the space between the carbon layers are relevant in the electronic structure in general and superconductivity in particular in GIC, where the interlayer states of the pristine graphite, located above the Fermi level, strongly hybridize with the valence states of the intercalant atoms and the electronic bands responsible for the hybridized states descend to intersect the Fermi level. In solid picene, we have found that, while there exist interlayer states (where we define a layer as the array of molecules in the ab plane) with amplitudes well away from the layers, they sit in energy 1.5 eV above the conduction band bottom (Fig.2 (b)). If the dopants are accommodated within the layer of the solid picene as suggested above, the dopant atoms are then expected to have little hybridization with the interlayer states of the solid picene.

To determine whether intralayer insertion of K atoms is energetically more favorable than interlayer insertion, we have prepared initial geometries by locating a K atom on each of 3×3 grid points on the ab plane of the pristine picene, from which we started optimization. The energies of the intralayer insertions are found to be about 0.2 eV lower in energy per molecule than those of the interlayer insertions. We have therefore performed an optimization for K₃picene starting from an initial geometry where six K atoms are put in between the molecules (four above the end benzene rings and two above the center rings) within the layer with the lattice parameters set to those observed for K_{2.9}picene. The angle between the molecular planes obtained is now obtuse, 114°, which is due to the expanded a and shrunk b axes. Analysis of the resultant structure shows that the smaller volume of the unit cell and the presence of the dopants cause a larger distortion of the planar shape of the molecules than in the pristine case. The electronic band structure is shown in Figure 4 (b) (solid lines). They significantly differ from those in the pristine picene, so that the rigid band assumption is invalid. In Fig.4 (b) we have also displayed (dashed lines) the result for a geometry with the K atoms removed but the other atoms and the doping level fixed. While the two results differ to some extent, which implies that potassium-states are hybridized in the conduction bands, a stronger origin of the non-rigidity against doping comes from the difference in the molecular orientations, which dominates the overlaps of the π MO's. The Bloch states near the Fermi level, however, turn out to have still the character of the LUMO and LUMO+1, while interlayer states were not found near the Fermi level.

The Fermi surface for the doped system in Fig.4 (c) consists of multiple surfaces, which contain warped, interwoven one-dimensional (planar) surfaces and a three-dimensional (pocket) surface, some of which involve transfers between MO's via K atoms.

To summarize we have obtained the band structure of solid picene with curious features. The geometry optimized here is not a decisive one, since there can be many metastable geometries, which was also the case with solid pentacene, ²² and further investigations should be needed. Future problems include how the features of the Fermi surface will have implications on mechanisms of superconductivity.

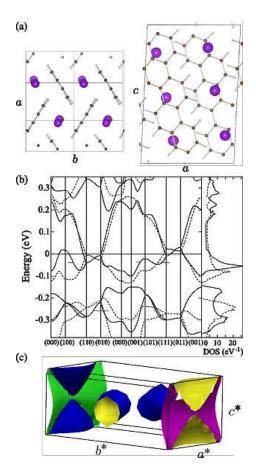


Fig. 4. (a) Optimized K₃picene geometry with large balls representing K atoms. (b) Band structure (solid curves with E=0 corresponding to E_F), along with that for the K-removed geometry for comparison (dashed curves with E=0 put equal to the position of E_F when we shift it to the situation for x=3). (c) The Fermi surface for the K₃picene.

We are indebted to Yoshihiro Kubozono for letting us know of the experimental results prior to publication. The present work is partially supported by the Next Generation Supercomputer Project, Nanoscience Program from MEXT, Japan, and by Scientific Research on Priority Areas "New Materials Science Using Regulated Nano Spaces" under Grant No. 19051016 from MEXT, Japan. The calculations were performed on the AIST Super Cluster at the Tsukuba Advanced Computing Center (TACC), AIST.

- N. B. Hannay, T. H. Geballe, B. T. Matthias, K. Andres, P. Schmidt and D. MacNair, Phys. Rev. Lett. 14 (1965) 225.
- T. E. Weller, M. Ellerby, S. S. Saxena, R. P. Smith and N. T. Skipper, Nature Phys. 1 (2005) 39; N. Emery, C. Hérold, M. d'Astuto, V. Garcia, Ch. Bellin, J. F. Marêché, P. Lagrange and G. Loupias, Phys. Rev. Lett. 95 (2005) 087003.
- D. Jerome, A. Mazaud, M. Ribault and K. Bechgaard, J. Phys. Lett. (Paris) 41 (1980) L95.
- H. Taniguchi, M. Miyashita, K. Uchiyama, K. Satoh, N. Mori, H. Okamoto, K. Miyagawa, K. Kanoda, M. Hedo and Y. Uwatoko, J. Phys. Soc. Jpn. 72 (2003), 468.
- A. F. Hebard, M. J. Rosseinsky, R. C. Haddon, D. W. Murphy, S. H. Glarum, T. T. M. Palstra, A. P. Ramirez and A. R. Kortan, Nature 350 (1991) 600.
- K. Tanigaki, T. W. Ebbesen, S. Saito, J. Mizuki, J. S. Tsai,
 Y. Kubo and S. Kuroshima, Nature 352 (1991) 222.
- T. T. M. Palstra, O. Zhou, Y. Iwasa, P. E. Sulewski, R. M. Fleming and B. R. Zegarski, Solid State Commun. 93 (1995) 327.
- R. Mitsuhashi, Y. Suzuki, Y. Yamanari, T. Kambe, N. Ikeda, H. Okamoto, A. Fujiwara, M. Yamaji, N. Kawasaki, Y. Maniwa, and Y. Kubozono, private communications.
- H. Okamoto, N. Kawasaki, Y. Kaji, Y. Kubozono, A. Fujiwara and M. Yamaji, J. Am. Chem. Soc. 130 (2008) 10470.
- 10) N. Kawasaki, Y. Kubozono, H. Okamoto, A. Fujiwara and M. Yamaji, Appl. Phys. Lett. 94 (2009) 043310.
- A. De, R. Ghosh, S. Roychowdhury and P. Roychowdhury, Acta Cryst. C 41 (1985) 907.
- P. E. Blöchl, Phys. Rev. B 50 (1994) 17953; G. Kresse and D. Joubert, *ibid.* 59 (1999) 1758.
- 13) http://www.qmas.jp/
- 14) K. Hummer and C. Ambrosch-Draxl, Phys. Rev. B 72 (2005), 205205.
- N. Marzari and D. Vanderbilt, Phys. Rev. B 56 (1997) 12847;
 I. Souza, N. Marzari and D. Vanderbilt, Phys. Rev. B 65 (2001) 035109.
- J. Yamauchi, M. Tsukada, S. Watanabe and O. Sugino, Phys. Rev. B 54 (1996) 5586.
- 17) N. Troullier and J. L. Martins, Phys. Rev. B **43** (1991) 1993.
- 18) Relevance of multiple LUMO's is reminiscent of Ni(tmdt)₂ [H. Seo, S. Ishibashi, Y. Okano, H. Kobayashi, A. Kobayashi, H. Fukuyama and K. Terakura, J. Phys. Soc. Jpn. 77 (2008) 023714], but this system is a single-component π-d system.
- T. Minakata, I. Nagoya and M. Ozaki, J. Appl. Phys. **69** (1991)
 7354; T. Minakata, H. Imai and M. Ozaki, J. Appl. Phys. **72** (1992) 4178; T. Ito, T. Mitani, T. Takenobu and Y. Iwasa, J. Phys. Chem. Solids **65** (2004) 609; Y. Matsuo, T. Suzuki, Y. Yokoi and S. Ikehata, J. Phys. Chem. Solids **65** (2004) 619; Y. Matsuo, S. Sakai and S. Ikehata, Phys. Lett. A **321** (2004) 62; B. Fang, H. Zhou and I. Homma, Appl. Phys. Lett. **86** (2005) 261909.
- A. Koma, K. Miki, H. Suematsu, T. Ohno and H. Kamimura, Phys. Rev. B 34 (1986) 2434.
- G. Csányi, P. B. Littlewood, A. H. Nevidomskyy, C. J. Pickard and B. D. Simons, Nature Phys. 1 (2005) 42.
- 22) R. G. D. Valle, E. Venuti and A. Brillante, J. Chem. Phys. 118 (2003) 807.